

# Determination of Mercury in Milk Powder by HG-AFS

Mercury is measured in milk powder by means of  $\text{H}_2\text{O}_2\text{-HNO}_3$  extraction followed by atomic fluorescence measurement.

### 1. Materials and methods of equipment and reagents

#### 1.1 equipment and reagents

AI3300 Atomic fluorescence spectroscopy;

Water bath with heater;

Adjustable micro pipettes (1mL, 100ul, 50ul);

Volumetric flask (10mL);

Concentrated  $\text{HNO}_3$ ;

30%  $\text{H}_2\text{O}_2$ ;

Reserved mercury standards solution ( $1000\mu\text{g}/\text{m}_\text{L} = 1000\text{ppm}$ ),

working mercury standards (0, 0.2, 0.5, 1.0, 2.0 ppb);

$\text{SnCl}_2$  solution:

Add 100gram  $\text{SnCl}_2$  in 100 mL concentrated HCl, water bath to totally dissolved, add water to 1000 mL.

High pure argon (>99.99%)

High pure distilled water.

### 2 Method

Accurately weight 1.00g milk powder in a 50 ml beaker, add a small amount of distilled water to moisten, add 1mL  $\text{H}_2\text{O}_2$  and wait 5 min, then add 5mL  $\text{HNO}_3$ , mix well, wait 20 min until the reaction stops. Place the beaker in a water bath and heat for 1 hour. Move the flask away from the water bath and let it cool down. Transfer all contents to a 10mL volumetric flask; add distilled water to make the total volume 10mL. At the same time, prepare the blank solution in the same way.

### 3. Instrument parameters

Carrier gas	300mL/min
Shield gas	800mL/min
HCL current	30mA
PMT voltage	360V
Integration time	6 s
Pump speed	40 r/min
Reducing reagent solution	10% $\text{SnCl}_2$ in 10% HCl (V/V)

### 4. Results

This method gives:

Detection limit: 0.3ppb,

Recovery rate: 95~110%

Relative standard deviation: 2~5%

